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# Structure Reports Online

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## Ethyl 3,5-bis(allyloxy)-4-bromobenzoate

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#### **Key indicators**

Single-crystal X-ray study  $T=120~\mathrm{K}$  Mean  $\sigma(\mathrm{C-C})=0.006~\mathrm{\mathring{A}}$  R factor = 0.052 wR factor = 0.084 Data-to-parameter ratio = 19.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

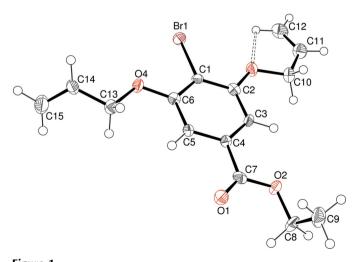
The asymmetric molecular conformation of the title compound,  $C_{15}H_{17}BrO_4$ , may be be influenced by an intramolecular  $C-H\cdots O$  interaction. The molecules form  $\pi-\pi$  stacks in the crystal structure.

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#### Comment

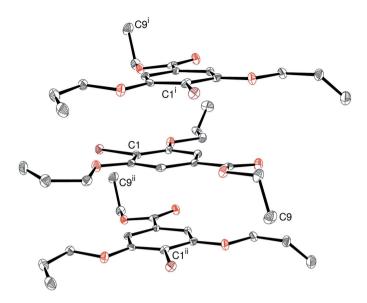
The title compound, (I) (Fig. 1), was prepared as part of our studies to determine the philicity of aryl radicals by competitive cyclization reactions (Kirsop *et al.*, 2004).

Compound (I) possesses normal geometrical parameters. The dihedral angle between the mean plane of the C1–C6 benzene ring and the plane of the C7/O1/O2 group is 6.0 (5)°. The two  $-O-CH_2-CH=CH_2$  side chains have very different conformations (Fig. 1), which may be attributable, at least in part, to an intramolecular C12 $-H12A\cdots$ O3 interaction (Table 1). The molecules form  $\pi$ – $\pi$  stacks in the crystal structure (Fig. 2), with alternating centroid-to-centroid



The molecular structure of (I), showing 50% displacement ellipsoids for non-H atoms. The intramolecular  $C-H \cdot \cdot \cdot O$  interaction referred to in the *Comment* is indicated by a dashed line.

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**Figure 2** Part of a  $\pi$ - $\pi$  stacked column of molecules (30% displacement ellipsoids and H atoms omitted). [Symmetry codes: (i) x, -y, 1-z; (ii) x, 1-y, 1-z.]

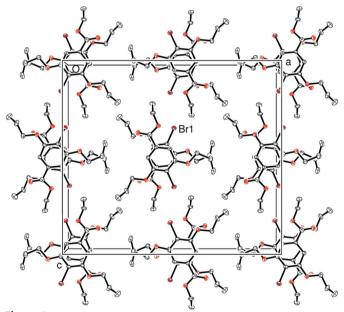


Figure 3
Unit-cell contents of (I), viewed down [010] (50% displacement ellipsoids and H atoms omitted).

separations between benzene rings  $[Cg \cdots Cg^i = 3.626 (2), Cg \cdots Cg^{ii} = 3.466 (2) \text{ Å; symmetry codes: (i) } x, -y, 1 - z; (ii) x, 1 - y, 1 - z]$ . The stacking interactions give rise to columns of molecules along [010] (Fig. 3).

## **Experimental**

4-Bromo-3,5-dihydroxybenzoic acid (6.8 g, 0.03 mol) was added to 100 ml of ethanol. Concentrated  $\rm H_2SO_4$  (1 ml) was added and the mixture was refluxed for 14 h. After cooling, the solvent was removed at reduced pressure to give a pale yellow oil. Diethyl ether (100 ml) was added and the mixture was neutralized by careful addition of a

saturated NaHCO<sub>3</sub> solution (100 ml). The mixture was transferred to a separating funnel and the product extracted with diethyl ether (4 × 100 ml). The combined extracts were dried over anhydrous MgSO<sub>4</sub> and evaporated under reduced pressure to give 4-bromo-3,5-dihydroxybenzoic acid ethyl ester as a white powder (7.5 g, 96%). Ethyl 4-bromo-3,5-dihydroxybenzoate (3.00 g, 0.011 mol), allyl bromide (1.30 g, 0.011 mol) and  $K_2CO_3$  (8.00 g, 0.0579 mol) were added to 100 ml of dry acetone. The mixture was stirred at room temperature under a nitrogen atmosphere for 14 h, then filtered and the solvent removed at reduced pressure to give a dark brown oil. Thin layer chromatography (4:1 hexane–ethyl acetate eluent) showed the title compound as a sharp spot at  $R_F = 0.52$ . The crude product was purified by flash column chromatography to yield a white powder (1.42 g, 38%). A sample of this powder was recrystallized from hot hexane to give translucent needles of (I) (m.p. 315–317 K).

#### Crystal data

 $\begin{array}{lll} C_{15} H_{17} Br O_4 & Z=8 \\ M_r=341.20 & D_x=1.483 \ {\rm Mg \ m^{-3}} \\ Orthorhombic, C222_1 & Mo \ K\alpha \ radiation \\ a=22.1421 \ (2) \ \mathring{A} & \mu=2.70 \ {\rm mm^{-1}} \\ b=7.0559 \ (13) \ \mathring{A} & T=120 \ (2) \ {\rm K} \\ c=19.5604 \ (11) \ \mathring{A} & {\rm Needle, colourless} \\ V=3056.0 \ (6) \ \mathring{A}^3 & 0.22 \times 0.04 \times 0.02 \ {\rm mm} \end{array}$ 

#### Data collection

 $\begin{array}{lll} \mbox{Nonius KappaCCD diffractometer} & 10933 \mbox{ measured reflections} \\ \mbox{$\omega$ and $\varphi$ scans} & 3495 \mbox{ independent reflections} \\ \mbox{$Absorption correction: multi-scan} & 2604 \mbox{ reflections with } I > 2\sigma(I) \\ \mbox{$T_{\rm min}$} & 0.588, T_{\rm max} = 0.948 & \theta_{\rm max} = 27.5^{\circ} \\ \end{array}$ 

### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.052$   $wR(F^2) = 0.084$  S = 1.01 3495 reflections 183 parameters H-atom parameters constrained 
$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0143P)^2] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &< 0.001 \\ \Delta\rho_{\text{max}} &= 0.47 \text{ e Å}^{-3} \\ \Delta\rho_{\text{min}} &= -0.53 \text{ e Å}^{-3} \\ \text{Absolute structure: Flack (1983),} \\ 1500 \text{ Friedel pairs} \\ \text{Flack parameter: } 0.106 \text{ (13)} \end{split}$$

**Table 1**Hydrogen-bond geometry (Å, °).

$D-\mathbf{H}\cdot\cdot\cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathbf{H}\cdot\cdot\cdot A$
C12—H12A···O3	0.95	2.39	2.715 (6)	100

H atoms were placed in idealized locations (C—H = 0.95–0.99 Å) and refined as riding atoms, with  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$  or  $1.5 U_{\rm eq}({\rm methyl~C})$ .

Data collection: COLLECT (Nonius, 1998); cell refinement: SCALEPACK (Otwinowski & Minor, 1997); data reduction: SCALEPACK and DENZO (Otwinowski & Minor, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

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# organic papers

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Acta Cryst. (2007). E63, o833–o835 Kirsop et al. • C<sub>15</sub>H<sub>17</sub>BrO<sub>4</sub> 0835